

## ULTREX SO 4

### 1 IDENTIFICATION

Product Code : 2300022

Recommended use of the chemical and restrictions on use: Industrial applications

Hubbard-Hall Inc.

563 South Leonard Street

Waterbury, CT 06708

Telephone : 203-756-5521

Fax number: 203-756-9017

Emergency Phone Number

CHEMTREC: 1 (800) 424-9300

International: 1 (703) 527-3887

### 2 HAZARDS IDENTIFICATION

Signal Word: DANGER

Hazard Category: Acute Toxicity-Oral Hazard Category 4

Eye Damage/Irritation Hazard Category 1

Skin Corrosion/Irritation Hazard Category 1A

Acute Toxicity-Inhalation Hazard Category 4

Specific Target Organ Toxicity (Single Exposure) Hazard Category 3

Corrosive to Metals Hazard Category 1

Acute Aquatic Toxicity-Category 2

Hazard Statements: Harmful if swallowed or inhaled.

Causes severe skin burns and eye damage.

May be corrosive to metals.

May cause respiratory irritation.

Toxic to aquatic life

Prevention: Wash skin thoroughly after handling.

Do not eat, drink or smoke when using this product.

Avoid breathing dust, fumes, gas, mist, vapors and sprays.

Wear rubber gloves, goggles and chemical protective clothing.

Use only outdoors or in well ventilated area.

Keep only in original container.

Response: If swallowed: Rinse mouth. Do NOT induce vomiting.

If on skin (or hair): Take off immediately all contaminated clothing Rinse skin with water/shower .

Wash contaminated clothing before reuse.

If inhaled: Remove person to fresh air and keep comfortable for breathing. Call poison center/doctor if you  
Immediately call poison center or doctor and explain the type of exposure to the chemical(s) and provide the  
Specific treatment - refer to poison center or doctor for advice.

If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to  
Absorb spillage to prevent material damage.

Storage: Store locked up.

Store in corrosive resistant high density polyethylene container.

Store in well ventilated place. Keep container tightly closed.

Disposal: Dispose of contents/container in accordance with local, regional, national, or international regulations.

### 3 COMPOSITION INFORMATION

Chemical Name	Common Name And Synonyms	CAS No. and other Unique identifiers	Concentration %
Hydrogen Chloride			



Relative Density:	1.01-1.02
Solubility (ies):	Complete in water
Partition Coefficient; n-octanol/water:	N/A
Auto-ignition Temperature:	N/A
Decomposition Temperature:	N/A
Viscosity:	N/A

## 10 STABILITY AND REACTIVITY

Chemical Stability:	Stable
Conditions to Avoid:	Extremely reactive. Avoid contact with metal surfaces and oxidizing agents.
Incompatible Materials:	Chemically stable when properly contained and handled. It is a strong mineral acid and reacts with many metals and metal oxides and hydroxides to form equivalent metal chloride. It reacts with zeolites and other silicious compounds to form Hydrosilicic Acid; it reacts with carbonates to form Carbon Dioxide and water. It is oxidized by Oxygen or electrolysis to form Chlorine, a lethal poisonous gas. It reacts with alkaline compounds to form neutral salt. It is a hydrolyzing agent for carbohydrates, esters and other compounds. Its reaction with most metals will produce Hydrogen, an explosive gas. Violent reactions will result with acetic anhydride, 2-aminoethanol, ammonia hydroxide, calcium phosphide, chlorosulfonic acid, ethylene diamine, ethylene imine, oleum, perchloric acid, beta propiolactone, propylene oxide, sodium hydroxide, uranium phosphide and vinyl acetate. This listing is not all inclusive.
Hazardous Decomposition Products:	Extreme heat may cause the product to decompose, producing toxic fumes which may include chlorine compounds.

## 11 TOXICOLOGICAL INFORMATION

Oral Administration:	Hydrochloric Acid-LD50:700 mg/kg (31.5%) rat
Inhalation:	Corrosive! Inhalation on vapors can cause coughing, choking, inflammation of the nose, throat, and upper respiratory tract, and in severe cases, pulmonary edema, circulatory failure, and death.
Dermal administration:	Hydrochloric Acid-LD50:>5010 mg/kg (31%) Rabbit
Short term exposure:	Hydrogen Chloride, both as a gas and in a solution such as Hydrochloric Acid, is a corrosive substance and can cause painful burns on contact with any part of the body or if taken internally. The mucous membranes of the eyes and upper respiratory tract are especially susceptible to the irritating effects of high atmospheric concentrations of Hydrogen Chloride. The gas or vapor is so penetrating and pungent that when high concentrations do occur, those exposed should immediately leave the contaminated area.
Long term exposure:	Long term exposure to concentrated vapors may cause erosion of the teeth. Long term exposure seldom due to corrosive properties of the acid.
Numerical measures of toxicity(such as toxicity measurement)	Hydrochloric Acid-Inhalation-LC50: 1.68mg/L, 1 h (rat) LC50: 4.72 mg/L, 1 h (rat)
Cancer Hazard:	Not listed by IARC, NTP, OSHA, ACGIH
Routes of Exposure	Eyes, Skin, Inhalation, Ingestion

## 12 ECOLOGICAL INFORMATION

Persistence and Degradability:	Not Available
Abiotic degradability:	No data available
Bioaccumulation potential:	No data available
Soil/Sediment Result:	Pronounced solubility and mobility

## 13 DISPOSAL CONSIDERATION

***Dispose of in accordance with local, state and federal regulations.***

14 TRANSPORT INFORMATION

UN Number: 1760  
UN Proper Shipping Name: CORROSIVE LIQUID N.O.S. (HYDROCHLORIC ACID),  
Transport Hazard Class (es): 8

15 REGULATORY INFORMATION