

Safety Data Sheet

Better Chemistry. Better Business

AQUAPURE™ I-200 PLUS

4/23/21

Product Name: AQUAPURE™ I-200 PLUS

Product Code :2601126

Recommended use of the chemical and restrictions on use:Industrial applications



Signal Word:
Hazard Category:

Hazard Statements:

Prevention:

Response:

Storage:

Chemical Name	Common Name And Synonyms	CAS No. and other Unique identifiers	Concentration %
Aluminum Sulfate	Alum	10043-01-3	~51.5%
Iron (II) Sulfate heptahydrate	Ferrous Sulfate Heptahydrate	7782-63-0	~46.5%
Sulfuric Acid	-	7664-93-9	~1%
Quarternary Ammonium Salt	-	31512-74-0	~1%

After Inhalation:

Remove from contaminated atmosphere. If breathing has ceased, clear the victim's airway and start mouth-to-mouth artificial respiration, which may be supplemented by the use of a bag-mask respirator, or manually triggered, oxygen supply capable of delivering 1 liter/second or more. If the victim is breathing, oxygen may be administered from a demand-type or continuous flow inhalor, preferably with a physician's advice. Contact a physician immediately.

After Skin Contact:

Immediately remove contaminated clothing under a safety shower. Flush all affected areas with large amounts of water for 15 minutes. DO NOT attempt to neutralize with chemical agents. Obtain medical advice.

After Eye Contact:

Immediately flush the eyes with large quantities of running water for 15 minutes. Hold the eyelids apart during the flushing to ensure rinsing of the entire surface of the eyelids with water. DO NOT attempt to neutralize with chemical agents. Obtain medical attention as soon as possible. Oils or ointments should not be used. Continue rinsing for an additional 15 minutes if the physician is not available.

After Ingestion:

DO NOT induce vomiting. Immediately give large quantities of water or milk, if available. If vomiting does occur, give fluids again. Never give anything by mouth to an unconscious person. Call a physician or the nearest Poison Control Center.

Most Important Symptoms/Effects

Delayed:

Will aggravate breathing disorders.

Suitable and Unsuitable extinguishing media:

If involved in a fire, use water spray. Neutralize with soda ash or slaked lime.

Specific hazards arising from the chemical:

Sulfur dioxide may be produced.

Carbon oxides may be produced.

Special protective equipment and precautions for firefighter

In the event of a fire, wear full protective clothing and NIOSH approved self contained breathing apparatus with full facepiece operated in the pressure demand or other positive pressure mode. Structural firefighter's protective clothing is ineffective for fires involving this material Stay away from ends of tanks. Cool tanks with water spray until well after fire is out.

Wear chemical resistant protective equipment and self contained breathing apparatus (SCBA).

**Personal Precautions,
Protective Equipment, &
Emergency Proc**

Prevent spilled product from drains, sewers, waterways and soil.

**Methods and Materials for
containment & cleaning up:**

Neutralize spill with soda ash or lime under good ventilation. For an interior (inside a closed space) spill be aware that the use of Soda Ash, Lime will evolve heat and carbon dioxide thus the need for ventilation.

If trained in accordance 29 CFR 1910.120, leaks should be stopped. Spills should be contained and cleaned immediately. Persons performing clean up work should wear adequate personal protective equipment and clothing. Spills and releases should be reported, if required, to the appropriate local, state and federal regulatory agencies.

Precautions for safe handling:

Use in well ventilated area.

Avoid breathing dust, fumes, gas, mist, vapors and sprays.

Wear rubber protective gloves, chemical protective clothing, eye protective goggles and face shield for face protection.

Wash hands thoroughly after handling.

Do not get in eyes, or on skin, or on clothing.

Eating, drinking and smoking in the work area is prohibited.

**Conditions for safe storage,
inc any incompatibilities:**

Keep container tightly closed.

Store locked up and away from incompatible chemicals.

Store in a well ventilated place. Keep cool .

Store in corrosive resistant container.

Do not allow material to freeze.

Name	Std.	TWA-8hrs	STEL - 15 min.
Aluminum Sulfate	ACGIH	2 mg/m ³	-
Iron (II) Sulfate Heptahydrate	ACGIH	1 mg/m ³ as Fe	-

ACGIH - American Control of Governmental Hygenists
OSHA - Occupational Safety and Health Administration

Appearance:	Brown liquid
Odor:	Very mild sweet
Odor Threshold:	N/A
PH:	0.8-1.2
Melting Point/Freezing Point:	N/A
Initial Boiling Point and Boiling Range:	N/A
Evaporation Rate:	N/A
Flammability (solid, gas):	N/A

Upper/Lower flammability or explosive limits:	N/A
Vapor Pressure:	N/A
Relative Density:	1.235-1.265
Solubility (ies):	Complete in water
Partition Coefficient; n-octanol/water:	N/A
Auto-ignition Temperature:	N/A
Decomposition Temperature:	N/A
Viscosity:	N/A
Chemical Stability:	Stable
Conditions to Avoid:	Extremely reactive. Avoid contact with metal surfaces and oxidizing agents.
Incompatible Materials:	Alkilies. Arsenic trioxide and sodium nitrate: spontaneously combustible mixture. Methyl isocynoacetate: may decompose explosively at 25 degrees
Hazardous Decomposition Products:	Sulfur Oxides may be formed. These are toxic and corrosive and are oxidizers. Sulfur trioxide is also a fire hazard. The loss of these gases leaves a caustic residue.
Oral Administration:	Aluminum Sulfate LD50(Rat)-1930 mg/kg
Oral Administration:	Iron (II) Sulfate Pentahydrate - LD50 (Mouse) - 1520 mg/kg
Oral Administration:	Sulfuric Acid-LD50-(Rat)-2140 mg/kg

**Sara Hazard
Classification**

Sulfuric Acid-Subject to reporting levels established by SARA Title III,Section 313

Proposition 65

WARNING! This product contains a chemical known in the State of California to cause cancer.
Strong inorganic mists containing sulfuric acid.

**TSCA Inventory
Status**

All components of this product are on the TSCA inventory or are exempt from TSCA inventory requirements .

Disclaimer:

The information is based on our knowledge to date but does not constitute an assurance of product properties and does not imply a legal contractual relationship.